



RN-6721

B. E. III (Sem. V) (Chem.) Examination

May / June - 2010

General Chemical Technology - I

Time : Hours]

[Total Marks : 100

Instructions :

(1)

नीचे दृष्टावेळ निशानीवाणी विगतो उत्तरवडी पर अवश्य लपवी. Fillup strictly the details of signs on your answer book.	Seat No. :
Name of the Examination :	<input type="text"/>
<input type="text" value="B. E. 3 (Sem. 5) (Chem.)"/>	<input type="text"/>
Name of the Subject :	<input type="text"/>
<input type="text" value="General Chemical Technology - 1"/>	<input type="text"/>
Subject Code No. : <input type="text" value="6"/> <input type="text" value="7"/> <input type="text" value="2"/> <input type="text" value="1"/>	<input type="text"/>
Section No. (1, 2,.....) : <input type="text" value="1&2"/>	<input type="text"/>
	Student's Signature

- (2) Answer each section **separately**.
(3) Figures to the right indicate full marks.
(4) Draw neat, labelled flow diagrams along with all the necessary chemical reaction wherever required.

SECTION - I

1 Attempt the following : 10

(a) Fill in the blanks :

- (i) The catalyst used in the manufacture of sulfuric acid by contact process is _____.
- (ii) 65% Oleum is prepared by distilling _____% oleum.
- (iii) The constituents of water gas are _____ and _____.
- (iv) Diaphragm electrolysis cell produces _____% NaOH solution.
- (v) The principal raw material for the production of chlorine gas by mercury electrolytic cell is _____ with _____.
- (vi) The chemical name of bleaching powder is _____.
- (vii) Glass industry required soda ash with _____ and _____.
- (viii) High alumina cement is manufactured by fusing _____ and _____.
- (ix) The catalyst used in the manufacture of nitric acid by ammonia oxidation process is _____.

- (x) Urea is formed in a low-stripping operation by dehydration of _____.
- (b) Describe the non-conventional sources of energy. 8
- 2** Attempt any **two** : 16
- (a) Discuss in detail the manufacture of soda ash by solvay process and also discuss the advantages and disadvantages of this process.
- (b) Explain waste water treatment by activated sludge process. Discuss treated effluent disposal management.
- (c) Describe mining of sulfur by frasc process.
- 3** Attempt any **two** : 16
- (a) Explain with a flowsheet any one method of recovery and treatment of natural gas and LPG.
- (b) What are phosphatic fertilizers? Discuss any one in detail.
- (c) What is cooking of coal? Describe fraction distillation of coal.

SECTION - II

- 4** (a) Attempt the following : 18
- (i) Mention the two most prominent types of detergent widely used today. 1
- (ii) TMP with respect to pulping stands for _____. 1
- (iii) Hydrated salt of magnesium sulfate is also known as _____. 1
- (iv) Write the chemical reactions involved in chemical recovery from black liquor during sulfate (Kraft) pulp process. 2
- (v) Enlist uses of molasses. 2
- (vi) During starch production from maize _____ is used in steep tank as Bacteriostatic. 1
- (vii) Enlist most widely used solvents for the extraction of vegetable oils. 1
- (viii) Define the term biodegradability with respect to surfactant. 1
- (b) Explain in detail major engineering problems involved in production of sugar manufacturing. 8

- 5** Attempt the following : (any **two**) **16**
- (a) Explain in detail hydrogenation of vegetable oil.
 - (b) Discuss various starch derivatives with their chemical structures.
 - (c) Draw a neat flow diagram of paper manufacturing from pulp stock.
- 6** Attempt the following : (any **four**) **4×4=16**
- (a) Write a short note : Additives for detergent.
 - (b) Draw only a neat flow diagram for production of dextrine.
 - (c) Enlist the major engineering problems associated with hydrogenation of vegetable oil and explain any one of them.
 - (d) Write a short notes on speciality papers.
 - (e) List advantages and disadvantages of continuous counter current process for fatty acid and soap production.
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